

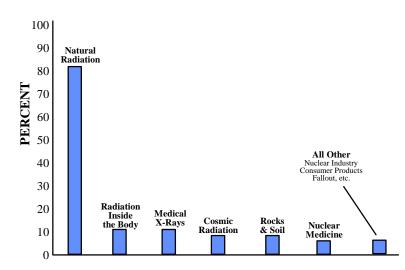
Radiation in the Environment



Radiation is a natural part of our environment. When our planet was formed, radiation was present — and radiation surrounds it still. Natural radiation showers down from the distant reaches of the cosmos and continuously radiates from the rocks, soil, and water on the Earth itself.

During the last century, mankind has discovered radiation, how to use it, and how to control it. As a result, some manmade radiation has been added to the natural amounts present in our environment.

Many materials — both natural and manmade — that we come into contact with in our everyday lives are radioactive. These materials are composed of atoms that release energetic particles or



waves as they change into more stable forms. These particles and waves are referred to as *radiation*, and their emission as *radioactivity*.

As the chart on the left shows, most environmental radiation (82%) is from natural sources. By far the largest source is radon, an odorless, colorless gas given off by natural radium in the Earth's crust. While radon has always been present in the environment, it's significance is better understood today. Manmade radiation — mostly from medical uses and consumer products — add about 18% to our total exposure.

Types of Ionizing Radiation

Radiation that has enough energy to disturb the electrical balance in the atoms of substances it passes through is called ionizing radiation. There are three basic forms of ionizing radiation.

Alpha

Alpha particles are the largest and slowest moving type of radiation. They are easily stopped by a sheet of paper or the skin Alpha particles can move through the air only a few inches before being stopped by air molecules. However, alpha radiation is dangerous to sensitive tissue inside the body.

Beta

Beta particles are much smaller and faster moving than alpha particles. Beta particles pass through paper and can travel in the air for about 10 feet. However, they can be stopped by thin shielding such as a sheet of aluminum foil.

Gamma

Gamma radiation is a type of electromagnetic wave that travels at the speed of light. It takes a thick shield of steel, lead, or concrete to stop gamma rays. X-rays and cosmic rays are similar to gamma radiation. X-rays are produced by manmade devices; cosmic rays reach Earth from outer space.

Units of Measure

Radiation can be measured in a variety of ways. Units of measure show either (1) the total amount of radioactivity present in a substance or (2) the level of radiation being given off.

The radioactivity of a substance is measured in terms of the number of transformations (changes into more stable forms) per unit of time. The *curie* is the standard unit for this measurement and is based on the amount of radioactivity contained in 1 gram of radium. The amounts of radioactivity that people normally work with are in the millicurie (onethousandth of a curie) or microcurie (one-millionth

of a curie) range. Levels of radioactivity in the environment are in the picocurie or pCi (one trillionth) range.

Levels of radiation are measured in various units. The level of gamma radiation in the air is measured by the *roentgen*. This is a relatively large unit, so measurements are often calculated in milliroentgens. Radiation absorbed by humans is measured in either rad or rem. The rem is the most descriptive because it measures the ability of the specific type of radiation to do damage to biological tissue. Typical measurements will often be in the millirem (*mrem*), or one-thousandth of a rem, range.

Cosmic Radiation

Cosmic radiation is high-energy gamma radiation that originates in outer space and filters through our atmosphere. Sea Level......26 mrem/year Atlanta (1,050 ft)...... 31 mrem/year Denver (5,300 ft).....50 mrem/year Minneapolis (815 ft......30 mrem/year Salt Lake (4,000 ft)......46 mrem/year

Terrestrial Radiation

Terrestrial sources are naturally radioactive elements in the soil and water such as uranium, radium, and thorium. Average levels of these elements are 1 pCi/g of soil.

U.S. (average)	26 mrem/year
Denver, CO	63 mrem/year
Nile Delta, Egypt	350 mrem/year
Paris, France	350 mrem/year
Kerala, India	400 mrem/year
McAlpe, Brazil	2,448 mrem/year
Pocos de Caldas,	
Brazil	7 000 mrem/year

Brazil.....7,000 mrem/year

Buildings

Many building materials, especially granite, contain naturally radio-active elements.

U.S. Capitol Bldg	85 mrem/year
Statue of Liberty	325 mrem/year
Grand Central Sta	525 mrem/year
The Vatican	800 mrem/vear

Radon

Radon levels in buildings vary, depending on geographic location, from 0.1 to 200 pCi/liter.

Average indoor radon levels....1.5 pCi/liter Occupational working limit...100.0 pCi/liter

Radiation in the **Environment**

Because the radioactivity of individual samples varies, the numbers given here are approximate or represent an average. They are shown to provide a perspective for concentrations and levels of radioactivity rather than dose.

> mrem = millirem pCi = picocurie

Food

Food contributes an average of 20 mrem/year, mostly from potassium-40, carbon-14, hydrogen-3, radium-226, and thorium-232.

Beer	390 pCi/liter
Tap Water	20 pCi/liter
Milk	1,400 pCi/liter
Salad Oil	4,900 pCi/liter
Whiskey	1,200 pCi/liter
Brazil Nuts	14 pCi/g
Bananas	3 pCi/g
Flour	0.14 pCi/g
Peanuts & Peanut Butter.	0.12 pCi/g
Tea	0.40 pCi/g

Medical Treatment

The exposures from medical diagnoses vary widely according to the required procedure, the equipment, and film used for X-rays, and the skill of the operator. Chest X-Ray...... 10 mrem Dental X-Ray.....100 mrem

Consumer Goods

Cigarettes-2 packs/day8,000 mrem/year
(polonium-210)
Color Television<1 mrem/year
Gas Lantern Mantle2 mrem/year
(thorium-232)
Highway Construction 4 mrem/year
Airplane Travel-39,000 ft0.5 mrem/year
(cosmic)
Natural Gas/Heating and
Cooking (radon-222)2 mrem/year
Phosphate Fertilizers4 mrem/year

Natural Radioactivity in Florida Phosphate Fertilizers (in pCi/gram)				
	xxxxx	xxxxx	xxxxx	
Ra-226	21.3	21.0	33.0	
U-238	20.1	58.0	6.0	
Th-230	18.9	48.0	13.0	
Th-232	0.6	1.3	0.3	

Porcelain Dentures	1,500 mrem/year
(uranium)	
Radioluminescent Cloc	k<1 mrem/year
(promethium-147)	
Smoke Detector	0.01 mrem/year
(americium-241)	

International Nuclear **Weapons Test Fallout**

From pre-1980 atmospheric tests (average for a U.S. citizen.....1 mrem/year